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## Rotational and long range diffusion in a lithium amide–lithium borohydride mixture

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On-board hydrogen storage is still a challenge for fuel cell vehicles and other mobile applications. Complex hydrides, which contain ions such as  $\text{BH}_4^-$  and  $\text{NH}_2^-$ , have a high hydrogen capacity in combination with a low weight of the storage material. For example,  $\text{Li}_4\text{BH}_4(\text{NH}_2)_3$  contains 11.1 wt.% hydrogen and desorbs more than 10 wt% at 573-673 K. In previous studies the high desorption temperature was reduced with additives. To understand the chemical behaviour and atomic motions of  $\text{Li}_4\text{BH}_4(\text{NH}_2)_3$ , we present an in situ phase analysis and quasielastic neutron scattering (QENS) during heating.

In situ X-ray diffraction was measured up to 573 K at P02 (DESY) and QENS was taken at TOFTOF (MLZ) in the temperature range 300-514 K.  $\text{Li}_4\text{BH}_4(\text{NH}_2)_3$  melts at 494 K and during heating crystallization of a second phase was detected and identified as  $\text{LiNH}_2$ , which remained a crystalline residue in the melted material. From the quasielastic signal rotational and long range motions were analysed and assigned to  $\text{BH}_4^-$  and  $\text{NH}_2^-$  of  $\text{Li}_4\text{BH}_4(\text{NH}_2)_3$  and of the crystallized  $\text{LiNH}_2$  phase.

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