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Lithium ion containing block copolymer membrane for lithium ion microbatteries

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Motivation

Lithium ion batteries with solid state electrolyte

- Safer than liquid electrolyte
- High energy density
- Roll-to-roll processing possible
- Flexible form factor



Materials and structure

- PS-b-PEO
- Poly(styrene-block-ethylene oxide)
- Diblock copolymer (total $M_w = 91.5$ kg/mol)
- Mechanically stable glassy PS domains ($M_w = 30$ kg/mol)
- Lithium conducting PEO part ($M_w = 61.5 \text{ kg/mol}$) tends to crystallize [1]

ncreasing

- Lightweight
- No memory effect
- Ultra thin
- Integrating devices



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Impedance Spectroscopy

- Measurements for different Li/EO ratios as well as for different temperatures.
- Debye circuit with constant phase elements instead of ideal capacitors chosen to be a suitable equivalent circuit (see inset below). [2]



Nyquist plot for polymer membrane with Li/EO

Lithium salt LiTFSI

Bis(trifluoromethane) sulfoimide



cylinders Morphology of a block copolymer with different volume fractions f_{a} [6]

Sample Preparation

Li-polymer hybrid films with various Li/EO ratios were prepared via solution casting using THF (tetrahydrofuran) as solvent.

Sample between two mica

windows for SAXS/WAXS



impedance

All preparation steps were performed under inert gas in a glove box due to the hygroskopic nature of the LiTFSI salt.



Crystallinity (WAXS)

- Wide angle X-ray scattering (WAXS) measurements carried out to analyze the influence of Li salt and temperature on the crystallinity of the PEO blocks.
- At low temperatures, crystalline peaks at 1.3 and 1.6 Å⁻¹ are observed, which are assigned by crystals of PEO blocks, at high temperatures (>55°C), the PEO crystals melt. [5]
- With increasing salt concentration PEO crystallization is suppressed by the incorporated

ratio of 0.06 [1]

the Li/EO ratios at different temperatures. [1]

Results indicate an increasing conductivity with increasing temperature and LiTFSI content due to the hopping of Li ions within the polymer chain: [4]



- Film morphology (SAXS)
 - Small angle X-ray scattering (SAXS) measurements were carried out to analyze the inner morphology of the lithium-polymer hybrid films.
 - SAXS profiles were fitted using a model assuming lamellar morphology of the PS-b-PEO films.
 - Two main structural peaks are observed, which shift to lower q-values with the increase of temperature, indicating an expanded periodic distance
 - The increased domain spacing at high temperatures is due to the melting of PEO crystals can be seen the WAXS results.







Outlook

- Addition of TiO_2 nanoparticles (size: 2-8 nm) to the system to prevent PEO crystallization.
- Different PS-*b*-PEO polymer with larger molecular weight
- Solution casting
- Ultra thin films via spin coating
- All contents dissolved in the same solvent (THF and methanol)
- Porous titania nanostructured films as future electrode



SAXS heating profiles for different temperatures and Li/EO ratios [1]



Ultra thin solid state polymer electrolyte lithium ion batteries

q (Å⁻¹) q (Å⁻¹) WAXS heating profiles for Porous titania film based on PS-b-PEO block

copolymer electrolyte. Sample was made with 100nm Titania film with solution casting of BCP solution with lithium salt concentration of r = 0.08(*left*) and *r* = 0.15 (*right*) [1]

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