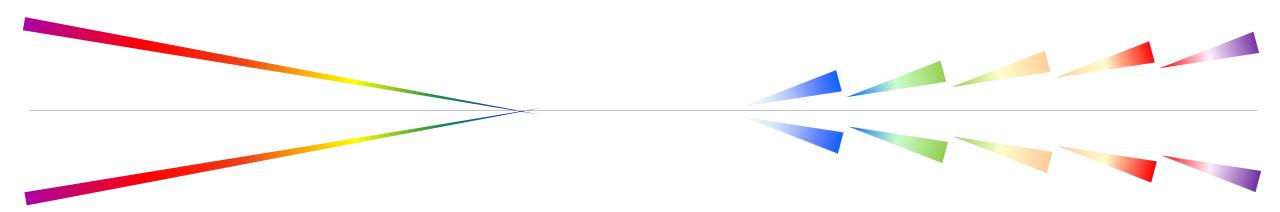


# Recent highlights and perspectives on chemical deuteration activities in CROSS



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Comprehensive Research Organization for Science and Society (CROSS)

#### J-PARC Workshop 2022 - Deuterium Science Entering a New Phase-

January 19th, 20th, 2023







Number of participant: 55 (young researcher is 12). Thanks a lot!

Next J-PARC Workshop will be held in October 2024 (tentative). We look forward to your participation (especially for young researchers)!

## **J-PARC MLF Deuteration Laboratory**

#### J-PARC MLF operates D-lab in cooperation with QST and CROSS.

#### **Development/management of deuteration facility (MLF)**

#### **Synthesis**

- Wet lab with fume hoods and clean bench
- High-pressure reactor
- Jar fermentor etc.

Characterization/analysis

- NMR (400 MHz)
- LC-MS/MS
- UV-Vis/FTIR spectroscopy
- QCM etc.



Development and expansion of facility



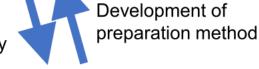
Development of preparation method

#### **Biological deuteration** (QST/MLF)

- Proteins and peptides
- DNA (Plasmid)
- E. coli cells



Development and expansion of facility



#### **Chemical deuteration** (CROSS)

Direct H-D exchange using H<sub>2</sub>/2-PrOH

- Carboxylic acids
- Ionic liquids
- Amides, etc.



## Current status of CROSS D-lab\_1

Since 2018, we have supplied some deuterated materials for neutron users.

#### Examples of chemical deuteration activities

- Synthesis of deuterated molecules for NR and inelastic neutron scattering for J-PARC MLF.
- Synthesis of deuterated molecules for SANS for JRR-3 (research reactor facility).
- Development of elemental analysis method for deuterated materials (in cooperation with JASCO Co.).
- Development of a new D<sub>2</sub>O recycling technique (in cooperation with FC Development Co.).



Number of collaborations on deuteration: 18

Number of deuterated molecules: 32

Number of publications: 8

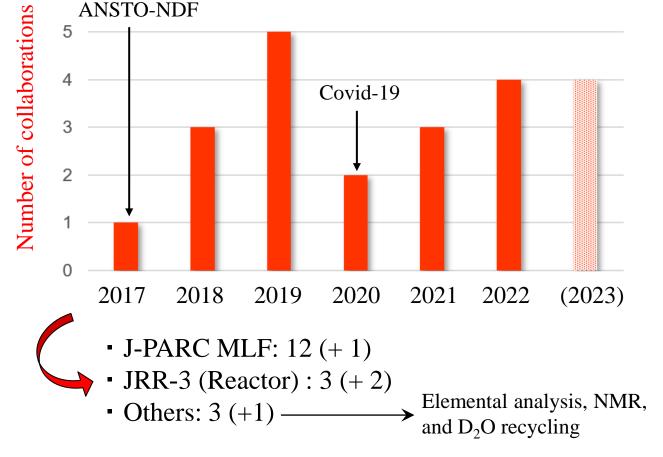
Ratio (molecules:publications): 32:8=4:1 **25%** 



It is necessary to increase the productivity of publications per deuterated molecules!!

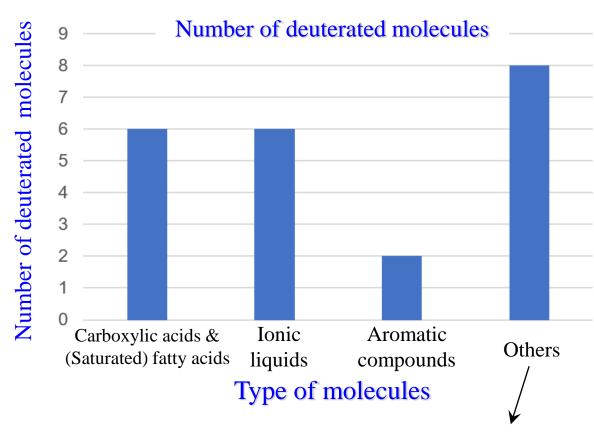
## Current status of CROSS D-lab\_2

Since 2018, we have supplied some deuterated materials for neutron users.



Number of collaborations

The number of acceptable collaborative research project is 4 or 5 per year.



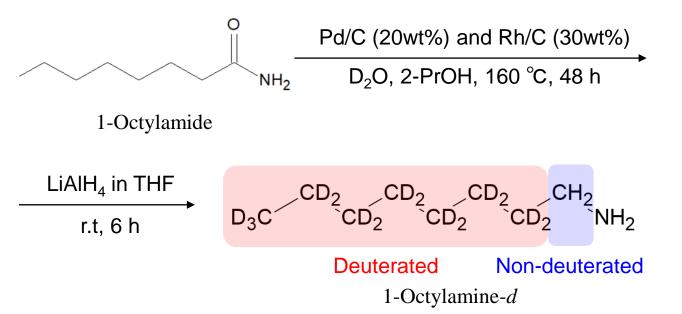
Alcohol, amine, amide, ketone, etc.

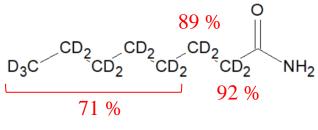
Carboxylic acids and ILs are in high demands. Recently, the demand for **deuterated amines** has been increasing.

### **Chemical Deuteration**

To increase chemical deuteration activities... 

A new synthetic route for deuterated amines





1-Octylamide-*d* 

→ Mean deuteration level: ~71 %

♦ New method from H. Sajiki's group\*

H<sub>3</sub>C 
$$N^+$$
(CH<sub>3</sub>)<sub>3</sub>  $N^+$ (CH<sub>3</sub>)<sub>3</sub>  $D_3$ C  $N^+$ (CD<sub>3</sub>)<sub>3</sub>  $D_3$ C  $N^+$ (CD<sub>3</sub>)<sub>3</sub>  $D_3$ C  $D_3$ C

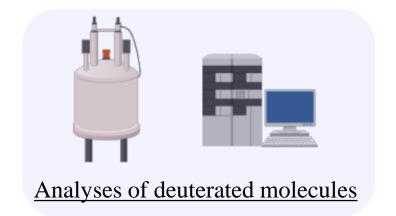
→ Mean deuteration level: ~50 %

\*Y. Sawama et al., Asian J. Org. Chem. In Press.

We will use these deuteration methods as needed.

# Analytical Techniques\_qNMR\_1

Development of new method for analyzing the chemical purity of deuterated compounds



Deuteration level analysis: <sup>1</sup>H, <sup>2</sup>H, <sup>13</sup>C-NMR, MS

Chemical purity analysis: ??

## Quantitative NMR (qNMR)\* analysis is an answer!?

\*The qNMR method was developed for determination of concentration and purity of drugs and medicines.

Development of <sup>1</sup>H and <sup>2</sup>H qNMR techniques is the best answer to analyze the isotopic and chemical purity of deuterated compounds!

**▶** This project is in progress.

# Analytical Techniques\_qNMR\_2

♦ ¹H/²H qNMR method is currently under development by



NMR standard reference: 
$$H_3C$$
  $GH_3$   $GH_3$ 

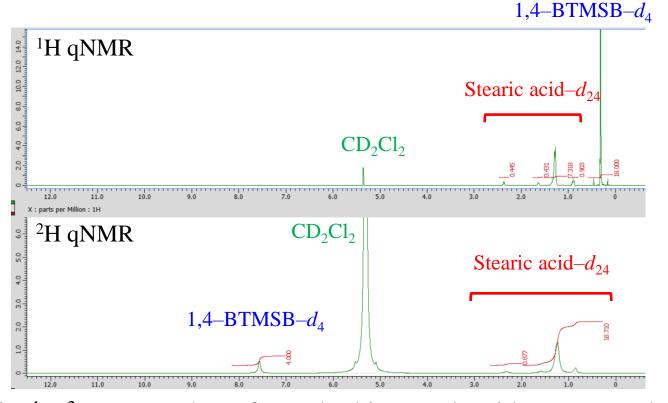


Fig.  ${}^{1}H/{}^{2}H$  qNMR data of round robin sample with NMR standard.

Device error, including operator error, is within  $\pm 1.0\%$ .

We will provide you with more details on our new qNMR method in the near future.

## Outcome\_Neutron Studies

Study on the Electric Double Layer Structure (J-PARC BL17)\* Aggregate structure of Pd-extractant (JRR-3 SANS-J)\*\*

\*K. Tamura et al, ECS Advances, 1, 020503 (2022).

\*M. Cyril et al, Solv. Extr. Ion Exch., 1-20 (2023).

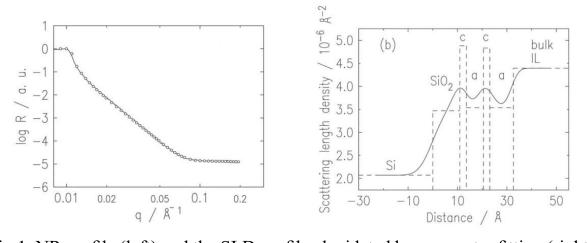


Fig. 1 NR profile (left) and the SLD profile elucidated by parameter fitting (right).

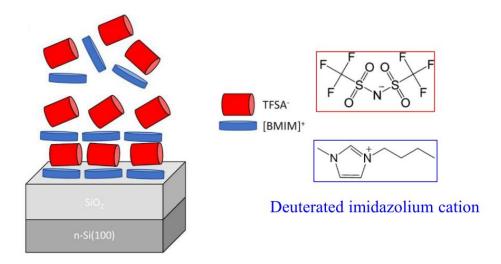
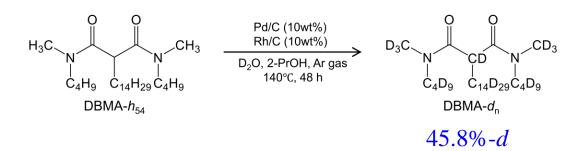


Fig.2 EDL structure obtained by parameter fitting.



Scheme 1 Direct deuteration reaction of DBMA.

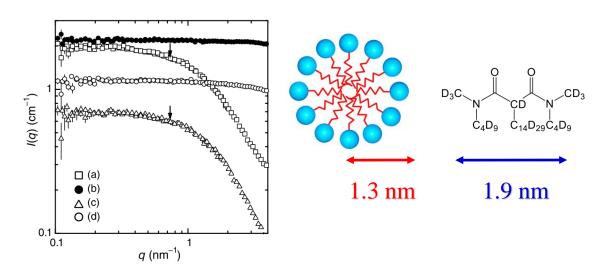
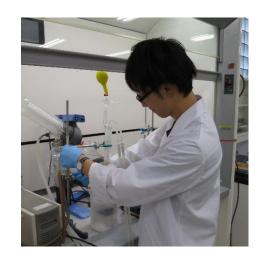


Fig.3 SANS profiles obtained for the four DBMA in *n*-heptane solutions.

# Summary

#### \* Perspectives on chemical deuteration activities

- 1) To increase the productivity of publications, easy and cost-effective synthesis method for deuterated amines is developing now.
- 2) The quantitative <sup>1</sup>H/<sup>2</sup>H NMR technique for the analysis of deuterated molecules will be well developed in the near future..
- 3) The number of acceptable collaborative research activities is 4 or 5 per year.
  - → For enhancing our deuteration activities, we need to find strong collaborators or employ a staff of deuteration chemist.
- ♦ For your information J-PARC&CROSS accept internship (or special research) students from various universities including overseas!





# Acknowledgement

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- JAEA/CROSS/QST
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