

Mullite-type materials: Introducing a new structure-type for the design of high-entropy oxides

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High-entropy oxides (HEO) present a new class of materials containing 5 or more metal cations in close to equal amounts in a single-phase solid solution. The high-entropy concept was first introduced in 2004 in the form of alloys [1], whereas the first HEO was presented in 2015 [2]. Until now, it has been assumed that HEOs have a strong tendency to crystallize in simple crystal structures e.g. rock salt, fluorite, perovskite and pyrochlore [3]. However, our study challenges this assumption by the first

successful synthesis of a HEO possessing a complex connectivity of polyhedral units in a mullite-type structure. The parent compounds $\text{Bi}_2\text{M}_4\text{O}_9$ ($\text{M} = \text{Al}^{3+}$, Ga^{3+} and Fe^{3+}) and $\text{REMn}_4\text{O}_{10}$ ($\text{RE} = \text{rare earth elements, Y and Bi}$) (Figure 1) show a variety of attractive properties such

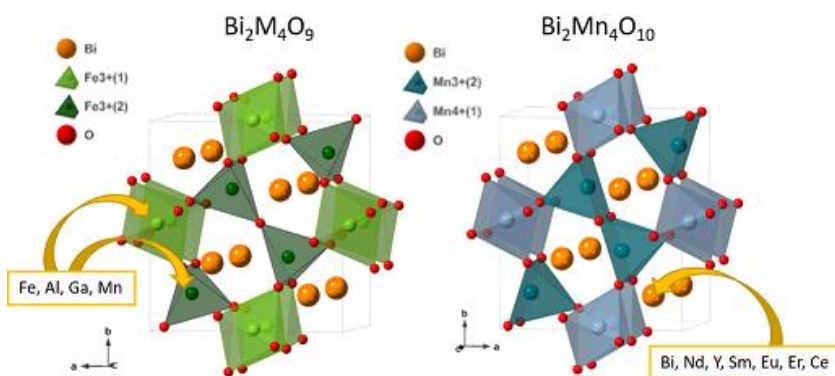


Fig.1 Crystal structures of the parent compounds $\text{Bi}_2\text{M}_4\text{O}_9$ and $\text{Bi}_2\text{Mn}_4\text{O}_{10}$

as multiferroicity and catalytic activity. The crystal structure is built up from edge-sharing MO_6 octahedral chains which are interconnected by double tetrahedral M_2O_7 units ($\text{Bi}_2\text{M}_4\text{O}_9$) or Mn_2O_8 double pyramids ($\text{REMn}_4\text{O}_{10}$).

In this work, we introduce 5 new HEOs with mullite-type structure $\text{Bi}_2(\text{AlGaFeMn})\text{O}_9$, $(\text{EuHE})_2\text{Mn}_4\text{O}_{10}$, $(\text{ErHE})_2\text{Mn}_4\text{O}_{10}$, $(\text{CeHE})_2\text{Mn}_4\text{O}_{10}$ ($\text{HE} = \text{Nd, Sm, Y, Bi}$) and $(\text{Nd, Sm, Y, Er, Eu})_2\text{Mn}_4\text{O}_{10}$. We show that the materials represent statistically mixed solid solutions using a combination of Neutron and X-ray powder diffraction with subsequent Rietveld analysis, X-ray total scattering and Pair distribution Function analysis, transmission electron microscopy, infrared and Raman spectroscopy. In addition, we follow their formation *in situ* by X-ray diffraction with a second scale time resolution. Surprisingly, all of them directly form out of an amorphous precursor without the formation of any other binary or ternary oxides. From a chemical point of view, the feasibility to crystallize structurally complex HEOs is fundamentally interesting and the targeted introduction of disorder into such a complex host lattice certainly bears great potential for the discovery of new physical phenomena.

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[2] Rost, C. M., Sacht, E., Borman, T., Moballegh, A., Dickey, E. C., Hou, D., Jones, J. L., Curtarolo, S., Maria, J.-P. Entropy-stabilized oxides. *Nat. Commun.*, 6, 8485 (2015).

[3] Witte, R., Sarkar, A., Kruk, R., Eggert, B., Brand, R. A., Wende, H., Hahn, H. High-entropy oxides: An emerging prospect for magnetic rare-earth transition metal perovskites. *Phys. Rev. Mater.* 3, 034406 (2019)