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Lattice modification and morphological control of halide substituted yqt-type zeolitic imidazolate frameworks Zn3mim5X, with X = Br, Cl, F or OH

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Zeolitic Imidazolate Frameworks (ZIFs) are a large subgroup of metal-organic frameworks, which has been studied for decades. The synthesis in aqueous solution using methylimidazolate (Hmim) and different zinc salts leads to the formation of Zn3mim5XH2O·nH2O with x = Br, Cl and α -Zn3mim5X with x = F, OH phases showing an yqt network. The imcorporation different halide and hydroxide anions into the network effects the de- and rehydration behavior, the spectral proteries and thermal expansion.

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