

## Structural diversity of carbonates containing CO<sub>4</sub> groups.

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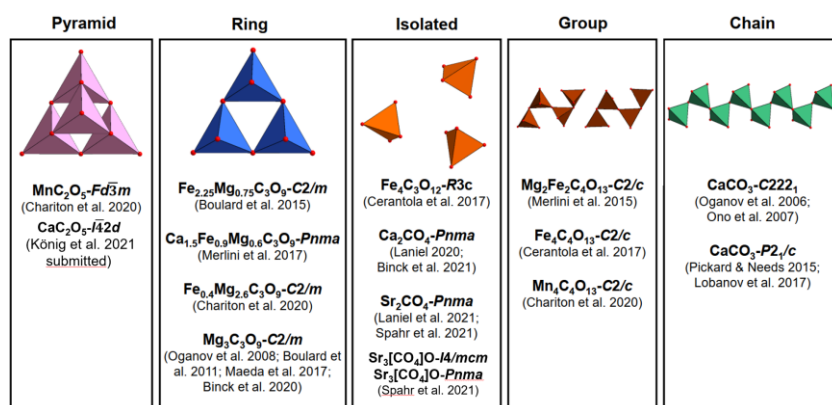
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The building blocks of ‘conventional’ carbonates such as calcite or magnesite are trigonal planar [CO<sub>3</sub>]<sup>2-</sup>-groups. These carbonates remain stable up to pressures of ~70 GPa. At higher pressures and high temperatures above ~2000 K the formation of [CO<sub>4</sub>]<sup>4-</sup>-groups was observed and explained by the formation of carbon with *sp*<sup>3</sup>-hybridized orbitals [1-3]. The experimental difficulties to achieve such extreme conditions hindered an extensive investigation of [CO<sub>4</sub>]-groups. In contrast to [CO<sub>4</sub>], other orthoanions [MO<sub>4</sub>] have been extensively investigated in the past. [SiO<sub>4</sub>]<sup>4-</sup>-tetrahedra are main building blocks in silicates and play a major role in crystallography/mineralogy. In addition to the [SiO<sub>4</sub>]<sup>4-</sup>-tetrahedra in silicates, further anions such as [PO<sub>4</sub>]<sup>3-</sup> (phosphates), [BO<sub>4</sub>]<sup>5-</sup> (borates), [NO<sub>3</sub>]<sup>3-</sup> (nitrates) or [SO<sub>4</sub>]<sup>2-</sup> (sulfates) are key-components in basic chemistry and are well-known building blocks of various minerals [4].

Recently, we demonstrated the synthesis of carbonates containing [CO<sub>4</sub>]-groups at moderately high pressures (20-30 GPa) by reacting carbonates with oxides or CO<sub>2</sub> [1-3]. These carbonates have different chemical compositions than the well-known ‘conventional’ carbonates (MeCO<sub>3</sub>) and are enriched either with the metal oxide or with CO<sub>2</sub> [1-3]. Some of them can even be recovered to ambient conditions [1,2]. This allowed us to investigate different structural aspects in great detail.

An interesting feature of the *sp*<sup>3</sup>-carbonates is that the [CO<sub>4</sub>]-groups may polymerize by corner-sharing. As a result, carbonates with isolated [CO<sub>4</sub>]-tetrahedra or carbonates with groups, rings, chains or pyramids can be formed (Fig. 1).



**Figure 1** Structural variety of carbonates containing [CO<sub>4</sub>]-tetrahedra. Further detailed references can be found in [1].

The structural variety is reminiscent of the structural variability of silicates [5]. In the study presented here we will give an overview of carbonates containing [CO<sub>4</sub>]-groups and will present crystal-chemical aspects of [CO<sub>4</sub>]-groups in comparison to [SiO<sub>4</sub>] and other [MO<sub>4</sub>] complex anions.

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