

Structural diversity of carbonates containing CO₄ groups.

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The building blocks of ‘conventional’ carbonates such as calcite or magnesite are trigonal planar [CO₃]²⁻-groups. These carbonates remain stable up to pressures of ~70 GPa. At higher pressures and high temperatures above ~2000 K the formation of [CO₄]⁴⁻-groups was observed and explained by the formation of carbon with *sp*³-hybridized orbitals [1-3]. The experimental difficulties to achieve such extreme conditions hindered an extensive investigation of [CO₄]-groups. In contrast to [CO₄], other orthoanions [MO₄] have been extensively investigated in the past. [SiO₄]⁴⁻-tetrahedra are main building blocks in silicates and play a major role in crystallography/mineralogy. In addition to the [SiO₄]⁴⁻-tetrahedra in silicates, further anions such as [PO₄]³⁻ (phosphates), [BO₄]⁵⁻ (borates), [NO₄]³⁻ (nitrates) or [SO₄]²⁻ (sulfates) are key-components in basic chemistry and are well-known building blocks of various minerals [4].

Recently, we demonstrated the synthesis of carbonates containing [CO₄]-groups at moderately high pressures (20-30 GPa) by reacting carbonates with oxides or CO₂ [1-3]. These carbonates have different chemical compositions than the well-known ‘conventional’ carbonates (MeCO₃) and are enriched either with the metal oxide or with CO₂ [1-3]. Some of them can even be recovered to ambient conditions [1,2]. This allowed us to investigate different structural aspects in great detail.

An interesting feature of the *sp*³-carbonates is that the [CO₄]-groups may polymerize by corner-sharing. As a result, carbonates with isolated [CO₄]-tetrahedra or carbonates with groups, rings, chains or pyramids can be formed (Fig. 1).

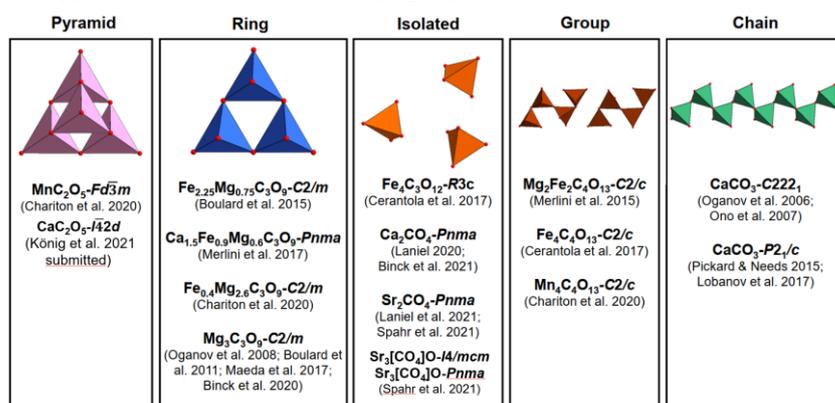


Figure 1 Structural variety of carbonates containing [CO₄]-tetrahedra. Further detailed references can be found in [1].

The structural variety is reminiscent of the structural variability of silicates [5]. In the study presented here we will give an overview of carbonates containing [CO₄]-groups and will present crystal-chemical aspects of [CO₄]-groups in comparison to [SiO₄] and other [MO₄] complex anions.

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[2] Spahr et al. *Inorg. Chem.*, 60, 5419 (2021)

[3] König et al. *Earth Space Chem.*, submitted (2021)

[4] Okrusch, M., Matthes, S., (Eds.) *B. Mineralogie*: Springer-Verlag: Berlin (2014)

[5] Liebau, F. *Structural Chemistry of Silicates*, Springer-Verlag: Berlin (1985)

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