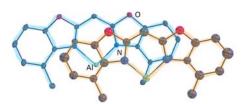
## Charge-Density Analysis of a disordered Aluminium Dihydride BOX Complex

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Today, single crystal X-ray diffraction (SCXRD) experiments are among the most valuable analytical methods for determining the three-dimensional structure of given crystalline substance. Charge-density analyses based on high-resolution datasets are an advanced SCXRD technique significantly more time consuming than the usual standard measurements. However, using the Hansen & Coppens Multipole Model [1], they allow for the modelling of electron density even in interatomic bonding regions. With Bader's Quantum Theory of Atoms in Molecules (QTAIM) [2], various structural properties of interest such as bonding characters and charge distribution can then be determined from a topological analysis of the model [3].

Due to an increased susceptibility to errors in the data, complications such as twinning or disorder are generally considered a dealbreaker in terms of charge-density analysis. Disordered structures come with a wide range of problems for charge-density refinement, first and foremost that many of the structural properties from charge-density analyses can often not be unambiguously determined in disordered parts [4]. In this work, we strive to investigate in how far a successful multipole refinement with subsequent Bader QTAIM analysis of the only slightly disordered



**Figure 1**: Structure of **1**. Main component highlighted in blue, second component highlighted in orange. Only isotropic displacement parameters were used for the second component Hydrogen atoms omitted for clarity.

compound  $AlH_2MeBOX$  [5] (1) can be achieved. Bis(benzoxazol-2-yl)methandid (BOX) ligands are a family of readily available ligands with easily customizable steric demand. Therefore, BOX ligands are a promising alternative to the ubiquitous NacNac ligands [6].

Compound  ${\bf 1}$  is the BOX analogue to an AlH<sub>2</sub>-NacNac complex showing catalytic activity, first reported in 2016 [7]. It contains two molecules in the asymmetric unit, both of which show a minor disorder with occupancies around 3% for the second position, respectively.

- [1] N. K. Hansen, P. Coppens, Testing Aspherical Atom Refinements on Small-Molecule Data Sets, *Acta Crystallogr. Sec. A* **1978**, *A34*, 909.
- [2] R. F. W. Bader, *Atoms in Molecules*. Clarendon Press, Oxford, New York **1990**.
- [3] U. Flierler, D. Stalke, More than just distances from Electron Density Studies , *Struct. Bond.* **2012**, *146*, 1.
- [4] K. Meindl, J. Henn, N. Kocher, D. Leusser, K. A. Zachariasse, G. M. Sheldrick, T. Koritsanszky, D. Stalke, Experimental Charge Density Studies of Disordered N-Phenylpyrrole and N-(4-Fluorophenyl)pyrrole, *Journal of Physical Chemistry A* **2009**, *113*, 9684.
- [5] J. Kretsch, A. Kreyenschmidt, T. Schillmöller, R. Herbst-Irmer, D. Stalke, Mixed Low-Valent Alanes from the Bis(4-methyl-benzoxazol-2-yl)methanid Ligand, *Inorg. Chem.* **2020**, *59*, 13690.
- [6] J. Kretsch, I. Koehne, M. Lõkov, I. Leito, D. Stalke, BOX Hounding NacNac: Varieties and Applications in Main Group Metal Coordination, *Eur. J. Inorg. Chem.* **2019**, 3258.
- [7] Z. Yang, M. Zhong, X. Ma, K. Nijesh, S. De, P. Parameswaran, H. W. Roesky, An Aluminum Dihydride working as a Catalyst in Hydroboration and Dehydrocoupling, *J. Am. Chem. Soc.* **2016**, *138*, 2548.