



DEVA Workshop: Neutron Activation Analysis

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MLZ is a cooperation between:











FRM II



<u>"classic in-core" NAA</u> \rightarrow instrument NAA

in-beam NAA → option at instrument PGAA

thermal, epithermal, fast neutrons

well-thermalized spectrum

- orders of magnitude higher flux than in-beam
- \rightarrow very low DLs (+)

only parallel **cold** neutrons

- large samples possible (+)
- attenuation is easy to calculate (+)
- much lower flux (-)



Core-near Positions + RCM Lab

Neutron Guide Hall West: NL4b









Neutron energy range used for elemental analysis:



Methods of chemical analysis with neutrons: INAA / NAA: instrumental neutron activation analysis incl. cNAA, ib-NAA,...
RNAA: radiochemical NAA
ENAA: epithermal NAA
FNAA: fast NAA
PGAA / PGNAA: prompt gamma(-ray neutron) activation analysis
PGAI-NT: Prompt Gamma-ray Activation Imaging and neutron tomography
NDP: neutron depth profiling
further methods: NRA, neutron-induced LSC, ...







Figure: https://nuclear-power.com







*I*₀: 0.55eV-1 MeV

 σ_0 : 0.025eV

1/v law

Capture probability / raction rate is highest for thermal and esp. cold neutrons.

Moderation to lower temperature makes sense in most cases.

> Threshold reactions do not occur in PGAA (with cold / thermal neutron beams). But they can disturb INAA and are used for FNAA.

Incident Energy (eV)

E-2

E-4

678 -6

78 \$+

Incident Neutron Energy (eV)





PGAA: mostly (n, γ) **NAA**: mostly (n, γ) + β -decay + γ







Chemical analysis with neutrons such as NAA enables a unique combination of advantages:

- ✓ Panoramic analyis with very low DLs for many elements.
- ✓ No (or not much) sample preparation / non- or low-destructive.
- ✓ High grade of matrix independency.
- ✓ High dynamic range.
- ✓ Real bulk method.
- ✓ NAA can be a so-called primary method of measurement (CCQM).
 → Its results can be deduced to fundamental constants using SI units (high traceability), also high reliability, high precision.





High dynamic range

PGAA

1 mg H together with 1 g Cl (10 mg water in 1 g CCl₄)

1 mg Cl together with 1 g H (1 mg Cl in 10 g water)



NAA

The HDR can be further increased by an irradiation-measurement plan adjusted to the half-lives of the activation products.

Parameters: irradiation time, decay time, counting time, detector-sample distance.





Sensitivity of INAA (typical geological material)



In-beam NAA: DL in g / INAA: DL in mg (ppt/ppq level possible!)

INAA = Instrumental Neutron Activation Analysis (no chem. treatment)





Sensitivity of PGAA (typical geological material)



Complementary Methods!





Lichen checking air pollution (PGAA + ibNAA)

z	EI	м	m	unc	m	unc	m	ox.	m	unc	mol%	unc	w%	unc	
			meas	%	Bkg	%	net	st.	ΟΧ	%	morra	%	el/el	%	
1	Η	1.008	1.18E-2	0.4	2.44E-7	1.7	1.18E-2	1	0.11	0.4	48.6	2.2	6.17	4.	٦
5	В	10.81	1.14E-6	0.4	3.87E-9	0.9	1.13E-6	3	3.65E-6	0.4	4.35 ppm	2.2	5.92 ppm	4.	
6	С	12.01	0.107	3.1	3.50E-2	3.0	0.07	4	0.26	5.	25	4.	38	5.	
7	Ν	14.01	2.32E-3	2.1		0.0	2.32E-3	5	8.95E-3	2.1	0.69	3.0	1.21	5.	
13	ΑΙ	26.98	7.32E-4	2.1	1.38E-4	2.0	5.93E-4	3	1.12E-3	2.6	0.091	3.4	0.31	5.	
14	Si	28.09	1.85E-3	3.1		0.0	1.85E-3	4	3.95E-3	3.1	0.27	3.7	0.97	5.	
16	S	32.07	2.47E-4	2.6		0.0	2.47E-4	6	6.17E-4	2.6	320 ppm	3.3	0.129	5.	
17	CI	35.45	9.91E-5	1.7	9.27E-8	2.4	9.90E-5	-1	9.90E-5	1.7	116 ppm	2.7	0.052	4.	Ì
19	Κ	39.1	9.92E-4	1.7		0.0	9.92E-4	1	1.20E-3	1.7	0.105	2.7	0.52	4.	
22	Ti	47.87	3.50E-5	3.0		0.0	3.50E-5	4	5.84E-5	3.0	30 ppm	3.7	180 ppm	5.	
25	Mn	54.94	5.05E-6	7.		0.0	5.05E-6	3	7.25E-6	7.	3.8 ppm	8.	26 ppm	9.	
48	Cd	112.4	3.06E-8	4.0		0.0	3.06E-8	2	3.49E-8	4.0	0.0113ppm	5.	0.16 ppm	6 .	
62	Sm	150.4	3.73E-8	2.8		0.0	3.73E-8	3	4.33E-8	2.8	0.0103ppm	3.5	0.19 ppm	5.	
64	Gd	157.3	4.45E-8	6.		0.0	4.45E-8	3	5.13E-8	6.	0.012ppm	6.	0.23 ppm	7.	ل
		0	-												
8	0	16	0.096	7.		0.0	0.10	-2		7.	25	6.	50	4.	
		0													_
11	Na	22.99	5.22E-5	0.7		0.0	5.22E-5	1	7.03E-5	0.7	94 ppm	2.3	273 ppm	4.	
17	CI	35.45	1.02E-4	3.0		2.4	1.02E-4	-1	1.02E-4	3.0	120 ppm	3.7	0.053	5.	
20	Ca	40.08	5.54E-3	13.		0.0	5.54E-3	2	7.75E-3	13.	0.6	13.	2.9	13.	
21	Sc	44.96	3.36E-8	4.		0.0	3.36E-8	3	5.15E-8	4.	0.031ppm	5.	0.18 ppm	6 .	
23	V	50.94	6.64E-7	23.		0.0	6.64E-7	5	1.19E-6	23.	0.5 ppm	23.	3 ppm	23.	
25	Mn	54.94	5.06E-6	2.2		0.0	5.06E-6	3	7.28E-6	2.2	3.8 ppm	3.1	26 ppm	5.	
26	Fe	55.85	1.00E-4	15.		8.	1.00E-4	3	1.43E-4	15.	70 ppm	15.	0.05	15.	
27	Со	58.93	1.25E-7	12.		0.0	1.25E-7	2	1.59E-7	12.	0.09 ppm	13.	0.7 ppm	13.	
35	Br	79.9	2.00E-6	8.		0.0	2.00E-6	-1	2.00E-6	8.	1.0 ppm	9.	10 ppm	9.	
38	Sr	87.62	1.09E-5	8.		0.0	1.09E-5	2	1.29E-5	8.	5.2 ppm	8.	57 ppm	9.	
63	Eu	152	7.21E-9	8.		0.0	7.21E-9	3	8.35E-9	8.	0.0020ppm	9.	0.038ppm	9.	_
		0													_



-PGAA

stoichiometry

- NAA



Sample preparation:

- Very sensitive to contaminations!
- Precise weighing necessary (typical sample mass in mg or µg range).

= 5 s

3200

$$S = 1 - e^{-\lambda t_{act}}$$

short + long irradiation

A few activation products are present in more than one measurement \rightarrow "intrinsic control"!

Evaluation step 1: Peak fitting with Hyperlab!

NAA – Theory

PGAA

- Gamma energy range up to 12 MeV
- Complicated spectrum with up to 10³ peaks
- Peak shape Gauss-like
- Baseline decreasing towards high energies
- Poisson statistics
- Peak positions -> identifying the elements
- Peak areas-> determining quantities

NAA

- Gamma energy range up to 2 (4) MeV
- Spectrum with up to 10² peaks
- Peak shape Gauss-like
- Baseline decreasing towards high energies
- Non-Poisson statistics (if changing count rate)
- Peak positions -> identifying the elements
- Peak areas, half lives -> determining quantities

NAA – Theory

$$\frac{{}^{N_{\rm p}}/{}_{t_{\rm m}}}{{\binom{N_{\rm p}}{}_{t_{\rm m}}}^*} = \frac{w}{w^*} \cdot \frac{S \cdot D \cdot C}{S^* \cdot D^* \cdot C^*} \cdot \underbrace{\frac{M^*}{M} \cdot \frac{\theta}{\theta^*} \cdot \frac{\gamma}{\gamma^*} \cdot \frac{\sigma_0}{\sigma_0^*}}_{\eta_{\rm s}} \cdot \frac{f + Q_0}{f + Q_0^*} \cdot \frac{\varepsilon_{\rm p}}{\varepsilon_{\rm p}^*}$$

*k*₀: universal nuclear constant*: comperator

$$c_{x}(ppm) = \frac{\left[\frac{N_{p,x}}{t_{m} \cdot S \cdot D \cdot C \cdot W}\right]}{A_{sp,Au}} \cdot \frac{1}{k_{0,Au}(x)} \cdot \frac{f + Q_{0,Au}(\alpha)}{f + Q_{0,x}(\alpha)} \cdot \frac{\varepsilon_{p,Au}}{\varepsilon_{p,x}} \cdot 10^{6} \qquad \text{OR:} \qquad \frac{1}{C}$$

$$\frac{c_{\rm x}}{c_{\rm std}} = \frac{a_{\rm x}}{a_{\rm std}}$$

$$f = \frac{\Phi_{\rm th}}{\Phi_{\rm e}'}$$
$$Q_0 = \frac{I_0}{\sigma_{\rm th}'}$$
$$A_{\rm sp,x} = N_{\rm p,x}/t_{\rm m} \cdot S \cdot D \cdot C \cdot W$$

 α : epitermal shape factor

🔚 Kayzero for V	Vindows					
File Samples Mon	itors Library Repo	orts History S	OLCOI Archive	Tools Windov	W Help Strategies Constraints	
E REPORT - PE	AK DATA SAMPL	E ORDER (64	D) DEMO:1B		🖾 Spectrum: 2S7V04.SPE	
246,91 268,22 273,40 275,94 320,00 328,80 345,660 373,09 373,09 40,04 438,47 479,38 479,49479,49 479,49 479,49479,49 479,49 47	$\begin{array}{c} 1259\\ 3680\\ 24355\\ 7973\\ 54262\\ 11220\\ 4457\\ 146107\\ 1538110\\ 8194\\ 309\\ 8194\\ 3056\\ 35826\\ 102802\\ 102$	1144266 109017 101061 97147 63363 59309 5332207 49388 49388 22047 33450 223365 22828 22047 33450 223365 22828 220345 220345 220345 20045 179052 179054 17905	$\begin{array}{c} 38, \ 32\\ 13, \ 159\\ 12, \ 03\\ 5, \ 67, \ 73, \ 68\\ 31, \ 79, \ 78\\ 78, \ 69\\ 29, \ 324\\ 70, \ 80\\ 27, \ 554\\ 0, \ 80\\ 27, \ 554\\ 0, \ 80\\ 27, \ 554\\ 0, \ 80\\ 0, \ 386\\ $	(101) (101) </th <th>LFC-spectrum :257VD4.SPE Spectrum plot : DEMO, 1B, SCK7V:2 13099.5753</th> <th>DEC:M</th>	LFC-spectrum :257VD4.SPE Spectrum plot : DEMO, 1B, SCK7V:2 13099.5753	DEC:M

Evaluation step 2: Calculations with Kayzero!

Corrections for calculating the concentration:

- background (only special cases, e.g. Co-60)
- blank (if not re-packaged)
- gamma absorption (large samples)
- neutron self-absorption (large samples)
- burn-up (only long-time with high flux)
- dead-time
- true-coincidence correction (if sample is close to the detector)
- inteferences
- threshold reactions (at FRM II mostly not relavant)
- fission

Evaluation step 2: Calculations with Kayzero!

NAA – Theory

Christian Stieghorst

Close cooperation with reactor operation (V. Hutanu).

Acquisition of gamma-ray spectra

 $5 \times 10^{12} - 7 \times 10^{13} \text{ cm}^{-2} \text{ s}^{-1}$

capsule irradiation

f = very high

- 3 HPGe detectors with digital spectrometers (Lynx, Canberra)
- **Evaluation**: Hyperlab + Kayzero (using k_0 method)
- New standard protocol for a complete analysis:
 - Short irradiation (few mins) I. 2 countings within 30 mins
 - Long irradiation (~1h) П. 2 countings 3 and 10 days later
- **Throughput**: 2-4 samples/day (only weekdays)

the user system.

Irradiation

NAA

Position	<i>∳</i> th(1/cm²s)	<i>∮_{≅pi}</i> (1/cm²s)	ø₁(1/cm²s)	f (th/epi)
RPA1	3,6E+13	6,7E+09	2,0E+09	5300
RPA2	1,5E+13	3,2E+09	4,1E+08	4800
RPA3	4,8E+12	7,6E+08	7,2E+07	6400
RPA4	7,3E+13	2,4E+10	5,6E+11	3000
RPA5	3,9E+13	1,2E+10	5,0E+09	3400
RPA6	7,1E+12	1,2E+09	1,5E+08	5700
KBA1-1	1,3E+14	2,6E+11	3,9E+11	500
KBA1-2	9,3E+13	9,9E+10	2,0E+11	940
KBA2-1	1,1E+14	7,5E+10	2,1E+11	1500
KBA2-2	7,7E+13	3,9E+10	1,0E+11	2000
SDA1	1,2E+13	1,0E+09	1,5E+10	12000

Other facilities ~10⁸ cm⁻² s⁻¹

- HPGe detector
- ⁶Li-containing plastic (2.5 mm)
- Scintillator annulus (Compton and cosmic muon suppression) was Nal, now BGO
- 10 cm of lead
- 5mm boron rubber (40% B_4C)
- 5cm boron plastic (20% H₃BO₃)

Concentration profile of 3d-transition metals in Si-Ingot:

Si-Ingot G2 irradiated at BR2

Karches, B., Welter, K., Stieghorst, C. *et al.* Instrumental determination of phosphorus in silicon for photovoltaics by β spectroscopy: a new approach. *J Radioanal Nucl Chem* **311**, 541–548 (2017). https://doi.org/10.1007/s10967-016-5051-7

Thank you for your attention!

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