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## Low-temperature phase transitions in PrAlO3-SrTiO3 series

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Phase and structural behaviour of the continuous perovskite solid solution Pr<sub>1-x</sub>Sr<sub>x</sub>Al<sub>1x</sub>Ti<sub>x</sub>O<sub>3</sub> have been studied in the temperature range 20 -295 K by highresolution X-ray synchrotron powder diffraction. Superb characteristics of the beamline ID22@ESRF allows to detect either subtle spiting of the main perovskite maxima and/or appearance of weak superstructure reflections, thus proving diverse variants of perovskite structure existing in Pr<sub>1-x</sub>Sr<sub>x</sub>Al<sub>1x</sub>Ti<sub>x</sub>O<sub>3</sub> series at different compositions and temperatures. It was revealed that the samples with x = 0.1 and 0.2 undergo a sequence of structural phase transitions R-3c –Immb –I2/m, similar to those observed for the parent PrAlO<sub>3</sub> phase. These LT transitions in praseodymium aluminate are unique among all RAIO<sub>3</sub> perovskites and are considered to be caused by the electronic effects involving Pr<sup>3+</sup> ions, e.g. a coupling between Pr<sup>3+</sup> electronic states and phonons and/or cooperative Jahn-Teller effects. It was established that temperatures of both R-3c –Immb and  $Immb-I2/m\ transitions\ in\ Pr<sub>1-x</sub>Sr<sub>x</sub>Al<sub>1-x</sub>Ti<sub>x</sub>O<sub>3</sub>Al<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<sub>Ti<s$ series systematically decrease from 205 K and 151 K for PrAlO<sub>3</sub> to 170 K and 90 K for x = 0.2 sample. Quite different phase behaviour was observed in the SrTiO<sub>3</sub>-rich part of the system. Simultaneous aliovalent substitution of Sr<sup>2+</sup> and Ti<sup>4+</sup> species by 10 % Pr<sup>3+</sup> and Al<sup>3+</sup> ions increases the temperature of a Pm3m –I4/mcm transition from 105 K in SrTiO<sub>3</sub> to~250~K~in~Pr<sub>0.1</sub>Sr<sub>0.9</sub>Al<sub>0.1</sub>Ti<sub>0.9</sub>O<sub>3</sub>.

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